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Transport and injection of CO₂ with impurities- When is corrosion an issue

VII November Conference, Rio 2019

Emission 2017: 35 Gt CO₂

IEA CCS Roadmap (2 °C)



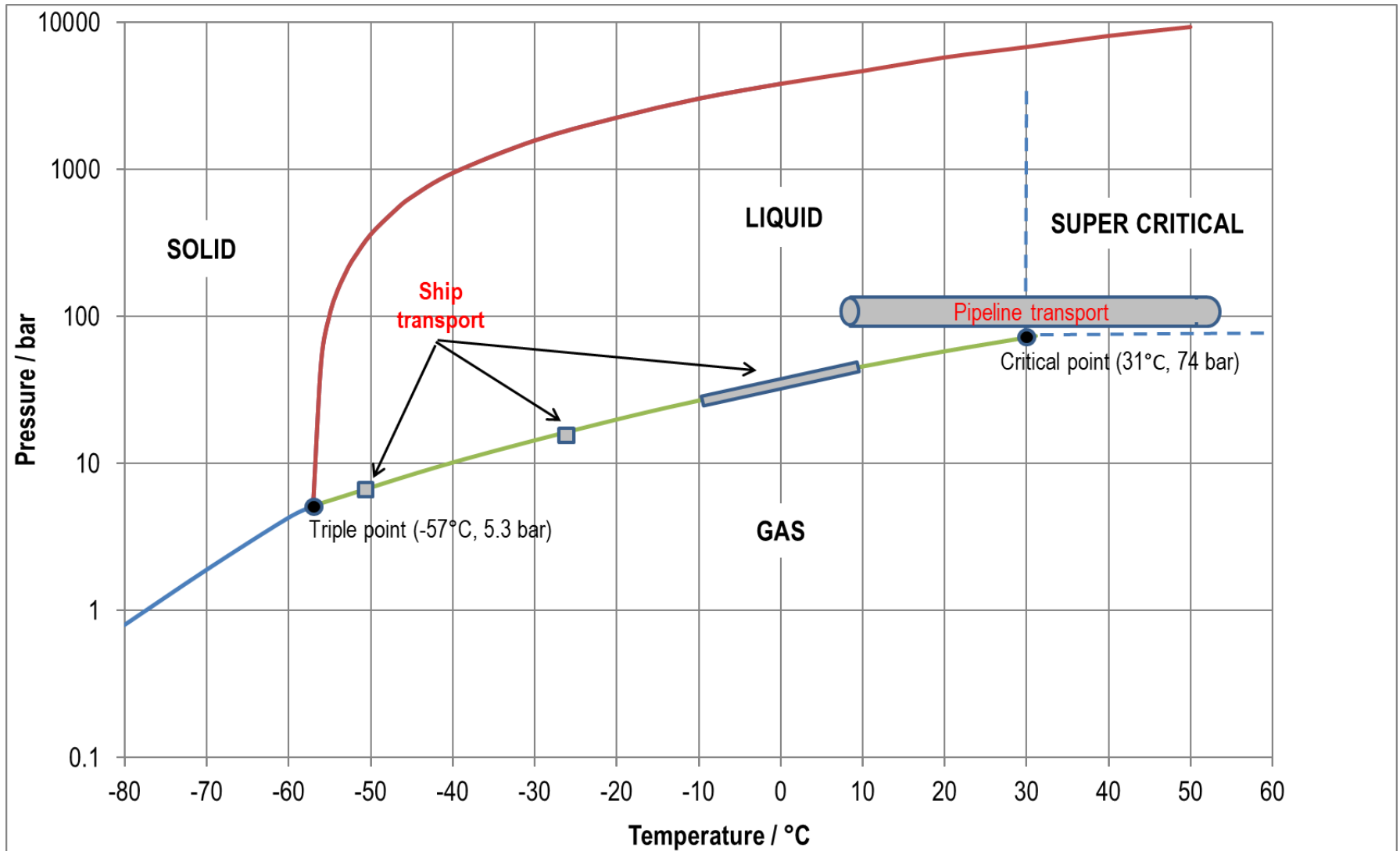
2050: transport and injection of 7-9 Gt CO₂



(2-3x)

12" pipelines ~ 3 000	}	Flow velocity ~1.5 m/s
20" pipelines ~ 1 000		
Ship (loading)~ 800 000	}	10 kton

Need to define the safe operation window before infrastructure and capture plants are designed and built!



CO₂ transport, State of the Art

- CO₂ injection for EOR > 35 years (USA)
- More than 100 installations, more than 5000 km pipeline
- **C-steel**: Good experience with clean and dry CO₂
- Reported corrosion when water accumulates
- CRA: "Wet" CO₂, Sleipner, short distance
- Thousands of papers/corrosion studies for pCO₂ < 20 bar
 - Few studies for pCO₂ > 50 bar
 - Few studies presenting data with flue gas impurities
- Ship transport , food grade quality
 - Yara (200 kton/year)

CCS CO₂ transport challenges vs. previous CO₂ transport experiences

- New impurities (H₂O, H₂S, O₂, CO, SO_x and NO_x) might be present
 - How much is acceptable?
 - When will cross chemical reaction take place?
 - When will aqueous phases form
 - When will corrosion become a problem?
- Reuse of existing oil and gas infrastructure (13% Cr, soft material.....)
- Complex network
 - Compatibility of streams (monitoring)
 - Many point sources, more upsets?
- Subsea: shut down, pigging, depressurization and upset remediation
- Ship transport, low temperature (-25 C)-solubility, corrosive phases
- Use of flexible flowlines and risers

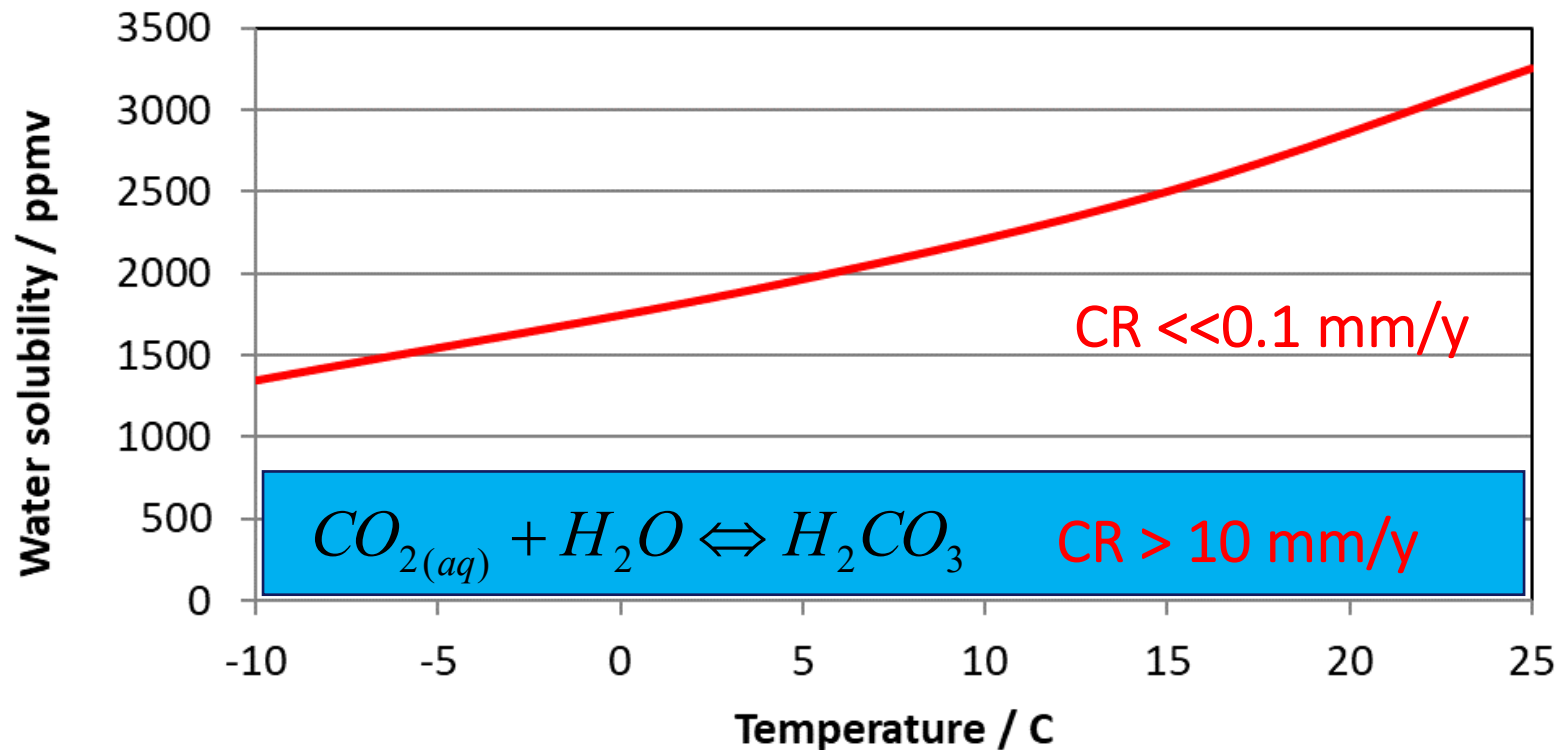
Safe operation window?

CO₂ composition recommendations/specifications

- Several recommendations, but lack of experimental verification
 - Based on: Corrosion issues, Health and safety, Reservoir/EOR
- The ISO transport standard (2016) gives no clear recommendation on CO₂ composition due to lack of data. *“The most up to date research should be consulted during pipeline design”*
- Does ship transport require stricter specifications than pipeline transport?

Comp. ppmv	Dynamis 2008	NETL* (for design)		Goldeneye/ Peterhead 2014 (2016)	Northern Light 2018
		2012	2013		
H ₂ O	500	730	500	50	30
H ₂ S	200	100	100	0.5	9
CO	2000	35	35	10	100
O ₂	<40000	40000	10	1 (5)	10
SO _x	100	100	100	10	10
NO _x	100	100	100	10	10

Water solubility in pure liquid CO₂

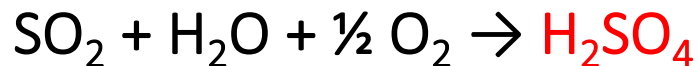
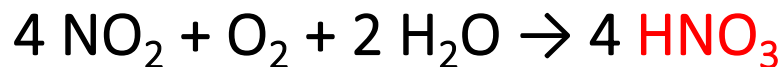


Impurities in CO₂ and possible reactions

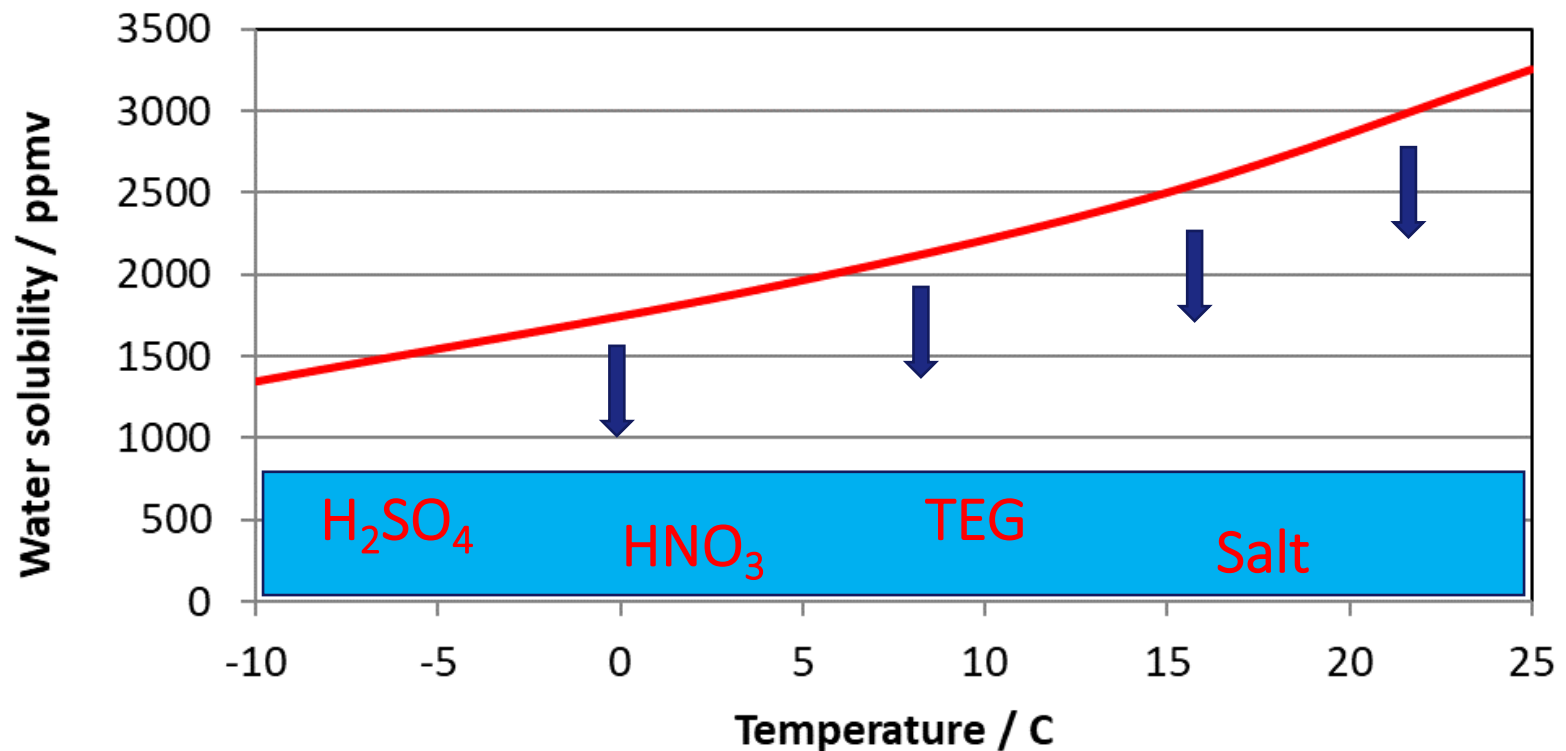
H ₂ O, ppmv	500
H ₂ S, ppmv	200
CO, ppmv	35
O ₂ , ppmv	40000
SO _x	100
NO _x	100

Solubility in pure CO₂: 1000-3000

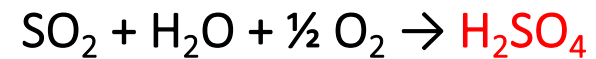
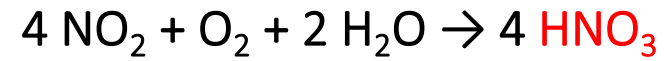
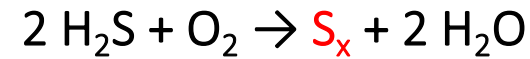
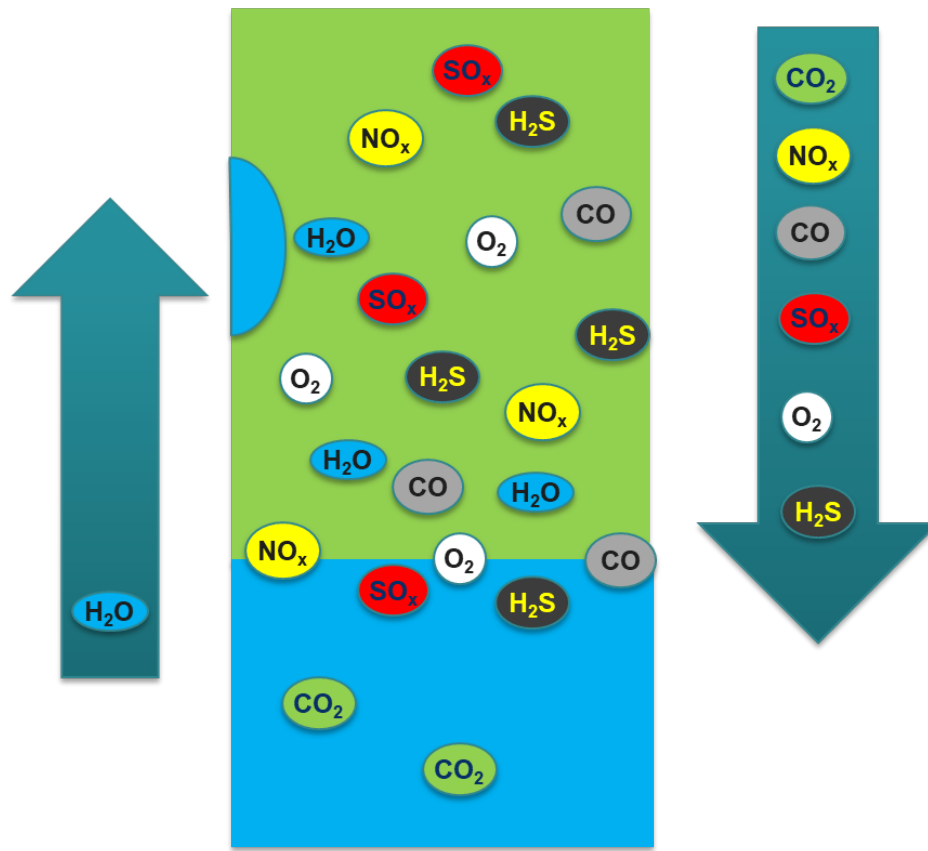
1.5 Mt/y
100 ppmv
100-200t/y



Water solubility in liquid CO₂ with impurities



Injection wells- backflow of formation water



$V_{\text{CO}_2}/V_{\text{water}}$ ratio

Testing of CO₂ specifications

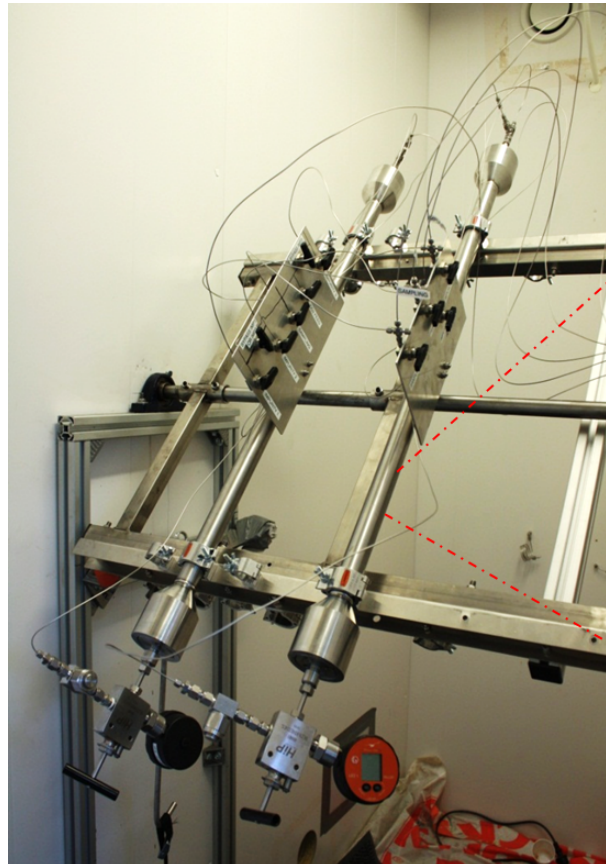
Comp. ppmv	Dynamis 2008	NETL* (for design)		Goldeneye/ Peterhead 2014 (2016)	Northern Light 2018	Exp 25 C 100 bar
		2012	2013			
H ₂ O	500	730	500	50	30	122/90
H ₂ S	200	100	100	0.5	9	130/36
CO	2000	35	35	10	100	0/0
O ₂	<40000	40000	10	1 (5)	10	275/70
SO _x	100	100	100	10	10	96/30
NO _x	100	100	100	10	10	69/32

	H ₂ O ppmv	H ₂ S ppmv	NO ₂ ppmv	NO ppmv	SO ₂ ppmv	O ₂ ppmv
IFE experiment. feed	122	130	96	0	69	275

1 liter autoclave
200 ppmv H₂O



ONE!
(< 0.1 ml)



8 cm

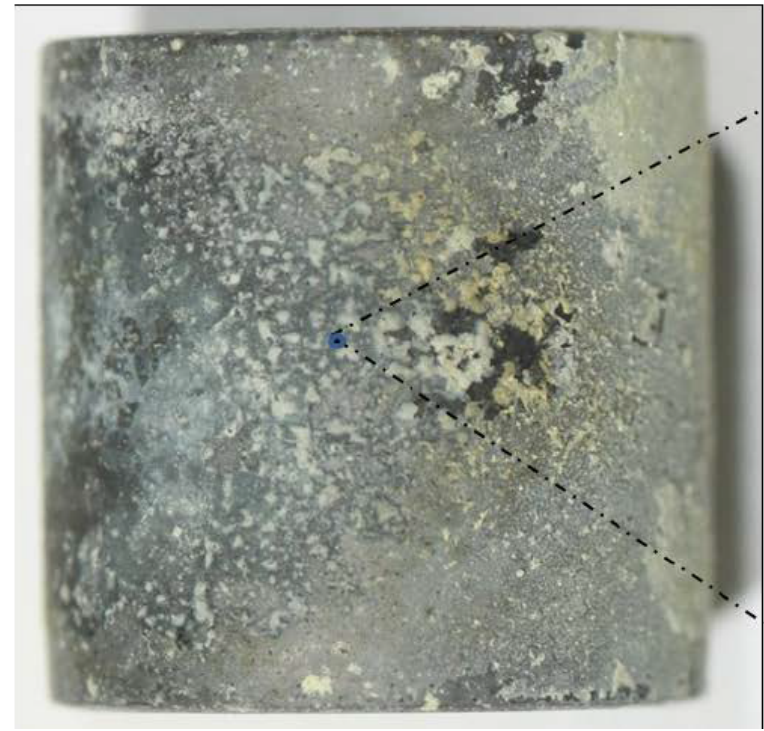


Sliding specimen rack

	H ₂ O ppmv	H ₂ S ppmv	NO ₂ ppmv	NO ppmv	SO ₂ ppmv	O ₂ ppmv
Feed	122	130	96	0	69	275
Vented CO ₂	<50	<10	<20	2-5	<30	Uncertain

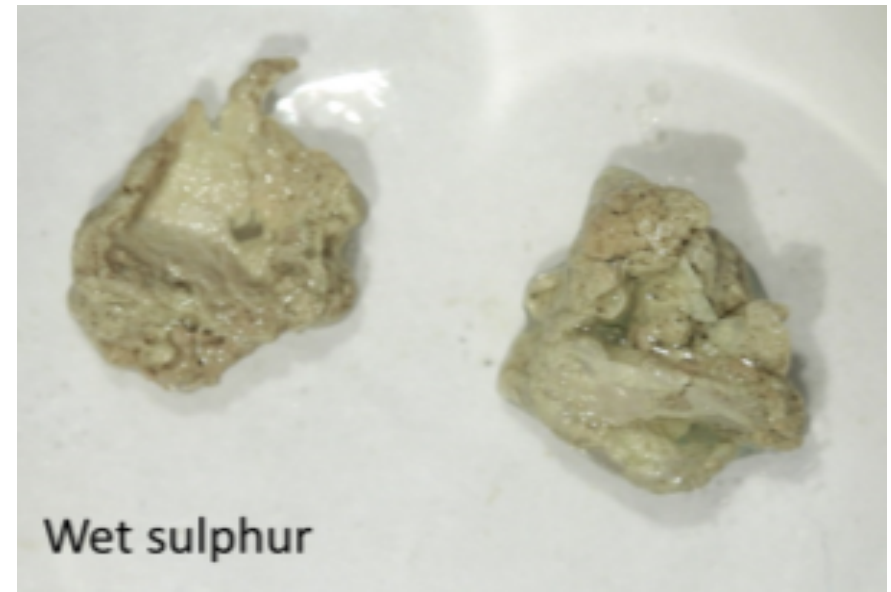
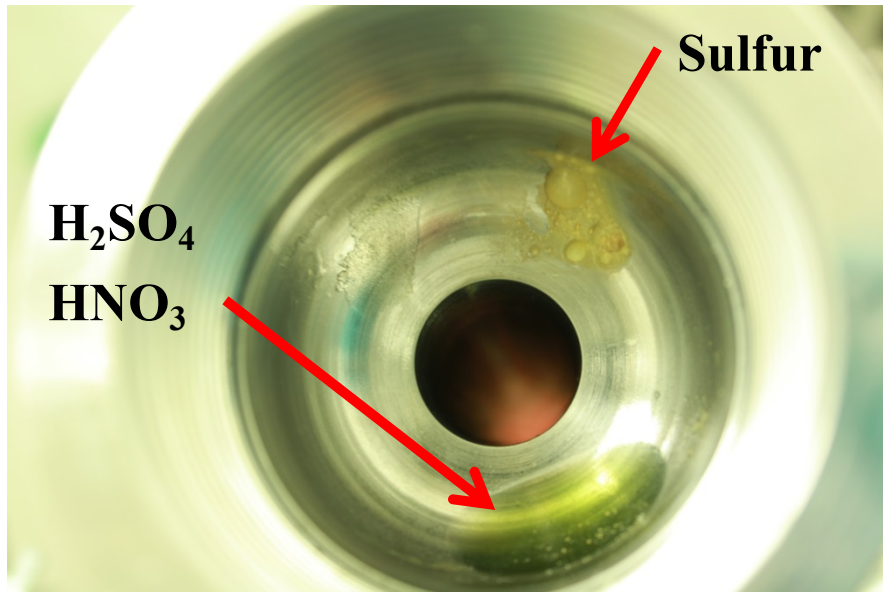


Not exposed



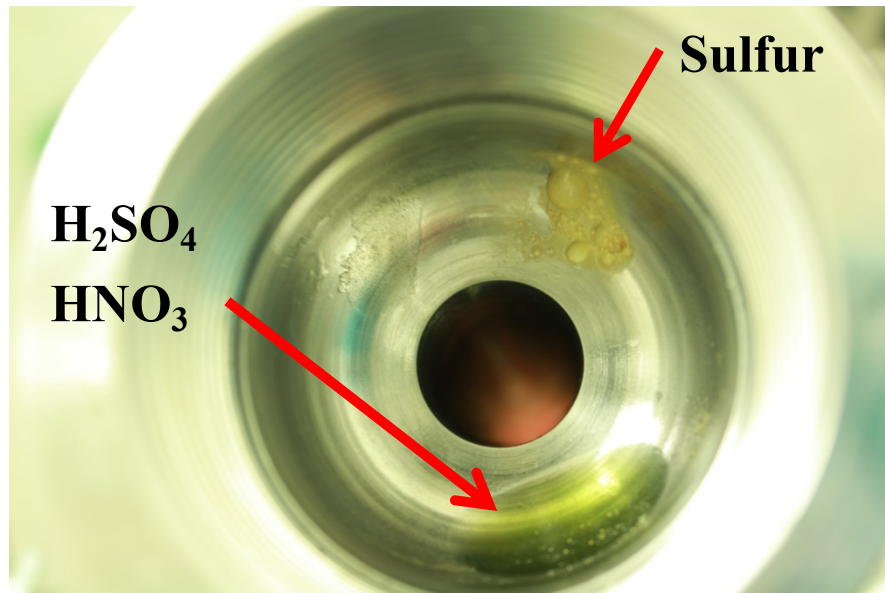
0.034 mm/year

	H ₂ O ppmv	H ₂ S ppmv	NO ₂ ppmv	NO ppmv	SO ₂ ppmv	O ₂ ppmv
Feed	122	130	96	0	69	275
Vented CO ₂	<50	<10	<20	2-5	<30	Uncertain



CO₂ 1.5 Mt/y, > 200t/y

	H ₂ O ppmv	H ₂ S ppmv	NO ₂ ppmv	NO ppmv	SO ₂ ppmv	O ₂ ppmv
Feed	122	130	96	0	69	275
Vented CO ₂	<50	<10	<20	2-5	<30	Uncertain



Is material selection a solution?

Conclusions/Challenges

- Free water and CO₂: high corrosion rates, $\gg 1$ mm/y
- No aqueous phase, low corrosion rates, $\ll 0.1$ mm/y
- Developing models/tools to determine/predict “safe operation window”
 - Formation of corrosive phases
 - Formation of solids (sulfur, corrosion product)
 - Accumulation of impurities (depressurization)
- Generate experimental data for validation of models and tools required to design and operate CCS systems efficiently and at minimum cost.